# DEPARTMENT OF SCHOOL EDUCATION <br> <br> Government JEE Coaching- 2019-20 

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## UNIT TEST- 7

Time: 60min
Marks: 180

## Instructions:

## 1) Answer all the questions

2) For Every correct answer Four marks will be given
3) For Every wrong answer One mark will be deducted

CHOOSE THE CORRECT ANSWER
45x4=180

1. There are two wires of same material and same length while the diameter of second wire is 2 times the diameter of second wire is 2 times the diameter of first wire, then ratio of extension produced in the wires by applying same load will be
1) $1: 1$
2) $2: 1$
3) $1: 2$
4) $4: 1$
2. The strain stress curves of three wires of different materials are shown in the figure. $P, Q$ and $R$ are the elastic limits of the wires. The figure shows that

1) Elasticity of wire $P$ is maximum
2) Elasticity of wire $Q$ is maximum
3) Teusile strength of $R$ is maximum
4) None of the above is true
3. A force $F$ is needed to break a copper wire having radius $R$. The force needed to break a copper wire of same length and radius 2 R will be
1) $\frac{F}{2}$
2) $F$
3) 4 F
4) $\frac{F}{4}$
4. A sample of a liquid has an initial volume of 1.5 L . The volume is reduced by 0.2 ml . The pressure increases by 140 KPa . What is the bulk modulus of the liquid?
1) $\frac{F}{2}$
2) $F$
3) 4 F
4) $\frac{F}{4}$
5. A uniform cube is subjected to volume compression. If each side is decreased by $1 \%$ then bulk strain is
1) 0.01
2) 0.06
3) 0.02
4) 0.03
6. The compressibility of a material is
1)Product of volume and its pressure
2) The change in pressure per unit always in volume strain
3) The fractional change in volume per unit
4) None of the above change in pressure
7. Force of 100 N each are applied in opposite direction on the upper and lower forces of a cube of side 20 cm . The upper face is shifted parallel to itself by 0.25 cm . If the side of the cube were 10 cm , then the displacement would be
1) 0.25 cm
2) 0.5 cm
3) 0.75 cm
4) 1 cm
8. Two wires of same diameter of the same material having the length 1 and $2 l$. If the force $F$ is applied on each, the ratio of the work done is the two wires will be
1) $1: 2$
2) $1: 4$
3) $2: 1$
4) $1: 1$
9. When an elastic material with young's modulus ' y ' is subjected to stretching stress 's' elastic energy per unit volume of the material is
1) $\frac{Y S}{2}$
2) $\frac{S^{2} Y}{2}$
3) $\frac{S^{2}}{2 Y}$
4) $\frac{S}{2 Y}$
10. If two equal and opposite deforming forces are applied parallel to the cross-sectional area of the cylinder as shown in the figure, there is a relative displacement between the opposite faces of the cylinder. The ratio of $\Delta x$ to L is known as

1)Longitudinal strain
2) Volumetric strain
3) Shearing strain
4) Poisson's ratio
11. Bulk modulus of water is $2 \times 10^{5} \mathrm{~N} / \mathrm{m}^{2}$. The change in pressure required to increase density of water by $0.1 \%$ is
1) $2 \mathrm{~N} / \mathrm{m}^{2}$
2) $2 \times 10^{8} \mathrm{~N} / \mathrm{m}^{2}$
3) $2 \times 10^{6} \mathrm{~N} / \mathrm{m}^{2}$
4) $2 \times 10^{4} \mathrm{~N} / \mathrm{m}^{2}$
12. The stress Versus strain graphs for wires two materials ' $A$ ' and ' $B$ ' are as shown in figure. If $Y_{A}$ and $Y_{B}$ are the young's modulii of the materials, then

1) $Y_{B}=2 Y_{A}$
2) $Y_{A}=Y_{B}$
3) $Y_{B}=3 Y_{A}$
4) $Y_{A}=3 Y_{B}$
13. The graph shown was obtained from experimental measurements of the period of oscillations ' T ' for different masses ' M ' placed in the scale pan on the lower end of the spring balance. The most likely reason for the line not passing through the origin is that the

1)Spring did not obey Hooke's law
2) amplitude of the Oscillations was toolarge
3) clock used needed regulating
4) mass of the pan was neglected
14. A work of $2 \times 10^{-2} \mathrm{~J}$ is done on a wire of length 50 cm and area of cross-section $0.5 \mathrm{~mm}^{2}$. If the young's modulous of the material of the wire is $2 \times 10^{10} \mathrm{~N} / \mathrm{m}^{2}$ then the wire must be 1)elongated to 50.1414 cm
2) contracted by 2.0 mm
3) stretched by 0.707 mm
4) none of these
15. A wire elongates by ' 1 ' mm when a load' $\mathbf{w}$ 'is hanged from it. If the wire goes over a pulley and two weights ' W ' each are hung at the two ends, the elongation of wire will be ( mm )
1) $\frac{l}{2}$
2) $l$
3) $2 \ell$
4) zero
16. Which one the following can not be prepared by WURTZ reaction
1) $\mathrm{CH}_{4}$
2) $\mathrm{C}_{2} \mathrm{H}_{6}$
3) $\mathrm{C}_{3} \mathrm{H}_{8}$
4) $\mathrm{C}_{4} \mathrm{H}_{10}$
17. The reactivity of hydrogen atom attached to carbon atom in the halogenation of alkane has the order
1) $3^{\circ}>1^{\circ}>2^{\circ}$
2) $2^{\circ}>1^{\circ}>3^{\circ}$
3) $3^{\circ}>2^{\circ}>1^{\circ}$
4) $1^{\circ}>2^{\circ}>3^{\circ}$
18. The molecules having dipole moments are
1)2,2 dimethy propane
2) trans-pent-2-ene
3)cis-hex-2-ene
3) Both $b \& c$
19. The number of benzylic hydrogen atoms in ethyl benzene is
1) 3
2) 5
3) 2
4) 7
20. Which of the compounds is not aromatic?
1) 


2)
3)
4)

21. Which of the following reaction is not shown by benzene

1) addition
2) oxidation
3) side chain substitution
4) $\pi$ complexformation
22. Nitration of benzene is
1)nucleophilic substitution
2) nucleophilic addition
3) electrophilic substitution
4) electrophilic addition
23. Which one of the following does not give an alkane?
1) Electrolysis of concentrated sodium acetate solution
2) Reaction of sodium acetate with soda lime
3) Reaction of $\mathrm{CH}_{3} \mathrm{I}$ with Na in ether
4) Reaction of ethyl chloride with alcoholic KOH
24. Cyclohexane a hydrocarbon floats on water because
1) it is immiscible \& lighter than water
2) it's density is low compared to water
3) it's a non polarsubstance
4) $\operatorname{both}(1) \&(2)$
25. 


is not converted to
 by

1) Wolffkishner reaction
2) Clemmensen reduction
3) red $\mathrm{P}+\mathrm{HI}$ at $200^{\circ} \mathrm{C}$
4) WURTZ reaction
26. 


(I) (II)
(III)
the correct order of reactinty towards addition reaction is

1) I $>$ III $>$ II
2) I $>$ II $>$ III
3) III $>$ II $>$ I
4) III $>$ I $>$ II
27. A compound A on ozyonolysis gives $\mathrm{CH}_{3} \mathrm{COCH}_{3}$ and HCHO. The IUPAC name of the compound is
1) 2 methyl propane
2) 2-methyl propane
3)1- methyl propene
3) 2 methyl propyne
28. One mole of symmetrical alkene on ozonolysis gives 2 moles of aldehyde having a molecular mass of 44 u . The alkene is
1) ethene
2) propene
3)1-butene
3) 2- butene
29. Propyne is formed by reaction of
1) $\mathrm{CH}_{3} \mathrm{Br}$ with sodium acetylide
2) $\mathrm{CH}_{3} \mathrm{I}$ with disodium acetylide
3) $\mathrm{CH}_{3} \mathrm{I}$ with acetone
4) $\mathrm{CH}_{3} \mathrm{I}$ with chloroethylene
30. Benzyl chloride $\left(\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2} \mathrm{Cl}\right)$ can be prepared from toluene by chloroethylene with I $-\mathrm{SO}_{2} \mathrm{Cl}_{2} \quad$ II- $\mathrm{SOCl}_{2} \quad$ III- $\mathrm{Cl}_{2} \quad$ IV -NaOCl
1) I \& III
2) $I \& I V$
3) II \& III
4) IV
31. The term independent of x in $\left(2 x^{2}-\frac{3}{x^{3}}\right)^{15}$ is
1) $\frac{9}{64}$
2) $\frac{8}{45}$
3) $\frac{64}{45}$
4) $\frac{45}{64}$
32. $5^{\text {th }}$ term of $\left(\sqrt{x}+\frac{a}{x}\right)^{12}$ is independent of x and it is 1980 . Then $\mathrm{a}=$
1) 2
2) $\sqrt{2}$
3) $\pm \sqrt{2}$
4) $\pm 2$
33. If the third term in the binomial expression if $(1+x)^{m}$ is $-\frac{1}{8} x^{2}$, then the rational value of $m$ is
1)2
2) $\frac{1}{2}$
3) 3
4) 4
34. If the third term is the expansion of $\left(\frac{1}{x}+x \log _{10} x\right)^{5}$ is 1 then $\mathrm{x}=$
1) 1
2) 10
3) 100
4) $\frac{1}{\sqrt{10}}$
35. The co efficient of $x^{3}$ in $\left[\sqrt{x^{5}}+\frac{3}{\sqrt{x^{3}}}\right]^{6}$ is
1) 0
2) 120
3) 420
4) 540
36. The coefficient of $x^{n}$ is $(1+x)^{n} \cdot\left(1+\frac{1}{x}\right)^{n}$ is
1) 0
2) 1
3) $2^{n}$
4) $2 \mathrm{nc}_{\mathrm{n}}$
37. The number of integral terms in the expansion of $(\sqrt{3}+\sqrt[8]{5})^{256}$ is
1)33
2) 34
3) 35
4) 32
38. In the binomial expansion of $(\mathrm{a}-\mathrm{b})^{\mathrm{n}}, \mathrm{n} \geq 5$, the sum of $5^{\text {th }}$ and $6^{\text {th }}$ terms is zero, then $\frac{a}{b}=$
1) $\frac{5}{n-4}$
2) $\frac{6}{n-5}$
3) $\frac{n-5}{6}$
4) $\frac{n-4}{5}$
39. If the coefficient of $\mathbf{x}^{7}$ and $\mathrm{x}^{8}$ is $\left(2+\frac{x}{3}\right)^{n}$ are equal, then $\mathrm{n}=$
1) 45
2) 55
3) 35
4) 27
40. If the coefficient of $r^{\text {th }},(\mathrm{r}+1)^{\text {th }}$ and $(\mathrm{r}+2)^{\text {th }}$ terms in the binomial expansion of $(1+y)^{\mathrm{m}}$ are in A.P, then $m$ \& $r$ satisfy the equation
1) $m^{2}-m(4 r-1)+4 r^{2}-2=0$
2) $m^{2}-m(4 r+1)+4 r^{2}+2=0$
3) $m^{2}-m(4 r+1)+4 r^{2}-2=0$
4) $m^{2}-m(4 r-1)+4 r^{2}+2=0$
41. If $28,56,70$ are the successive coefficient of $(1+x)^{n}$, then $n=$
1) 8
2) 9
3) 10
4) 11
42. The middle term in the expansion of $\left(x+\frac{1}{x}\right)^{10}$ is
1) $10 c_{6}$
2) $-10 c_{6}$
3) $-10 c_{5}$
4) $10 c_{5}$
43. The numerically greatest termof $(3 x+2 x y)^{11}$ when $x=2 / 3, y=3 / 4$ is
1) $10 c_{5} \times 486$
2) $-10 c_{5} X 486$
3) $11 \mathrm{c}_{5} \mathrm{X} 486$
4) $-11 \mathrm{c}_{5} \mathrm{X} 486$
44. If $(1+x)^{n}=C_{0} C_{1} x+C_{2} x^{2}+\cdots+C_{n} x^{n}$ then $\operatorname{Co}+2 \mathrm{C}_{1}+3 \mathrm{C}_{2}+-\cdots+(n+1) C_{n}=$
1) $2^{n}+n 2^{n-1}$
2) $2^{n-1}+n 2^{n}$
3) $2^{n}+(n+1) 2^{n}$
4) $2^{n-1}+(n-1) 2^{n}$
45. If $\mathrm{C}_{0}, \mathrm{C}_{1}, \mathrm{C}_{2} \ldots$. are the bionomical coefficient in the expansion of $(1+\mathrm{x})^{\mathrm{n}}$ then $\mathrm{C}_{0}+\mathrm{C}_{2}+\mathrm{C}_{4}+\mathrm{C}_{8}=$
1) $2^{7}$
2) 256
3) $2^{9}$
4) 258

## ANSWER KEY

| 1 | 4 | 16 | 1 | 31 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | 4 | 17 | 3 | 32 | 3 |
| 3 | 3 | 18 | 4 | 33 | 2 |
| 4 | 1 | 19 | 3 | 34 | 2 |
| 5 | 4 | 20 | 2 | 35 | 4 |
| 6 | 3 | 21 | 1 | 36 | 2 |
| 7 | 2 | 22 | 3 | 37 | 1 |
| 8 | 1 | 23 | 4 | 38 | 4 |
| 9 | 3 | 24 | 4 | 39 | 2 |
| 10 | 3 | 25 | 4 | 40 | 3 |
| 11 | 3 | 26 | 2 | 41 | 1 |
| 12 | 4 | 27 | 2 | 42 | 4 |
| 13 | 4 | 28 | 4 | 43 | 3 |
| 14 | 1 | 29 | 1 | 44 | 1 |
| 15 | 2 | 30 | 1 | 45 | 2 |

