

DEPARTMENT OF SCHOOL EDUCATION

Government JEE Coaching- 2019-20

UNIT TEST- 3

Time: 60 min

Marks: 180

Instructions:

- 1) Answer all the questions
- 2) For Every correct answer Four marks will be given
- 3) For Every wrong answer One mark will be deducted

CHOOSE THE CORRECT ANSWER

45x4=180

1. A body subjected to three concurrent forces is found to be in equilibrium. The resultant of any two forces
 - 1) Is equal to third force
 - 2) Is opposite to third force
 - 3) is collinear with the third force
 - 4) all of these
2. When forces \vec{F}_1 , \vec{F}_2 and \vec{F}_3 are acting on a particle of mass m such that \vec{F}_2 and \vec{F}_3 are mutually perpendicular, then the particle remains stationary. If the force \vec{F}_1 is now removed, then the magnitude of acceleration of particles
 - 1) $\frac{F_1}{m}$
 - 2) $\frac{F_2 F_3}{mF}$
 - 3) $\frac{F_2 - F_3}{m}$
 - 4) F_2/m
3. A weight w rests on a rough horizontal plane. If the angle of friction be θ , the least force that will move along the plane will be
 - 1) $w \cos \theta$
 - 2) $w \tan \theta$
 - 3) $w \cot \theta$
 - 4) $w \sin \theta$
4. A marble block of mass 2kg lying on ice when given velocity of 6 ms^{-1} is stopped by friction in 10 s . Then the coefficient of friction is
 - 1) 0.02
 - 2) 0.03
 - 3) 0.06
 - 4) 0.01
5. If reaction is R and coefficient of friction is μ , what is work done against friction in moving a body by distance d ?
 - 1) $\frac{\mu R d}{4}$
 - 2) $2\mu R d$
 - 3) $\mu R d$
 - 4) $\frac{\mu R d}{2}$
6. A 500kg car takes a round turn of radius 50m with a velocity of 36km/h. The centripetal force is
 - 1) 250N
 - 2) 750N
 - 3) 1000N
 - 4) 1200N
7. What will be the maximum speed of a car on a road turn of radius 30m if the coefficient of friction between the tyres and the road is 0.4 Taking $g=9.8 \text{ m/s}^2$
 - 1) 10.84 m/s
 - 2) 9.84 m/s
 - 3) 8.84 m/s
 - 4) 6.84 m/s
8. An unbanked curve has a radius of 60m. The maximum speed at which a car can make a turn if the coefficient of static friction is 0.75 is
 - 1) 2.1 m/s
 - 2) 14 m/s
 - 3) 21 m/s
 - 4) 7 m/s
9. A bus turns a corner on a slippery road at a constant speed of 12 m/s, If the coefficient of friction is 0.6, the minimum radius of the arc in metres in which the bus turns is
 - 1) 72m
 - 2) 24m
 - 3) 36m
 - 4) 9m
10. A stone tied to string is rotated with a uniform speed in a vertical plane. If mass of the stone is m , length of the string is r and linear speed of the stone is v , then tension in the string when the stone is at its lowest point is
 - 1) mg
 - 2) $\frac{mv^2}{r}$
 - 3) $\frac{mv^2}{r} - mg$
 - 4) $\frac{mv^2}{r} + mg$

- 11 A body of mass 4 kg is moving with momentum of 8 kg ms^{-1} . A force of 0.2 N acts on it in the direction of motion of the body for 10 s. The increase in kinetic energy is
 1) 10 J 2) 8.5 J 3) 4.5 J 4) 4 J
- 12 A particle acted upon by constant forces $4\hat{i} + \hat{j} - 3\hat{k}$ and $3\hat{i} + \hat{j} - \hat{k}$ is displaced from the point $\hat{i} + 2\hat{j} + 3\hat{k}$ to point $5\hat{i} + 4\hat{j} + \hat{k}$. The total work done by the forces in SI unit is
 1) 20 J 2) 49 J 3) 50 J 4) 30 J
- 13 When a long spring is stretched by 2 cm, its potential energy is V. If the spring is stretched by 10 cm, the potential energy in it will be
 1) 10 V 2) 25 V 3) $\frac{V}{5}$ 4) 5 V
- 14 Two spherical shaped solid masses undergo inelastic collision. Then
 1) Total kinetic energy is constant
 2) Total mechanical energy is not a constant
 3) Linear momentum will change
 4) Linear Momentum will not change
- 15 A mass of 5 kg is moving along a circular path of radius 1 m. If the mass with 300 revolutions per minute, its kinetic energy would be
 1) $250\pi^2$ 2) $100\pi^2 \text{ J}$ 3) $5\pi^2 \text{ J}$ 4) 0 J
- 16 Which type of bond is present in Xe molecule?
 1) Covalent 2) Ion dipole 3) Vander waal's 4) dipole- dipole
- 17 Weight of CH_4 in 9 L cylinder at 16 atm and 27° C is
 1) 0.92 g 2) 93.5 g 3) 3.84 g 4) 16 g
- 18 2 grams of hydrogen diffuse from a container in 10 minutes. How many grams of oxygen would diffuse through the same container in the same time under similar conditions?
 1) 0.5 g 2) 4 g 3) 6 g 4) 8 g
- 19 The kinetic energy of 4 moles of nitrogen gas at 127° C ---- cal. ($R = 2 \text{ cal mol}^{-1} \text{ K}^{-1}$)
 1) 4400 2) 3200 3) 4800 4) 1524
- 20 At high pressure the compressibility factor 'Z' is equal to
 1) Unity 2) $\frac{1-Pb}{RT}$ 3) $\frac{1+Pb}{RT}$ 4) zero
- 21 With rise in temperature, viscosity of a liquid
 1) increases 2) decreases 3) remains constant 4) may increase or decrease
- 22 The average kinetic energy of an ideal gas per molecule in SI unit at 25° C will be
 1) $6.13 \times 10^{-21} \text{ KJ}$ 2) $6.13 \times 10^{-21} \text{ J}$ 3) $6.13 \times 10^{-20} \text{ KJ}$ 4) $6.13 \times 10^{20} \text{ J}$
- 23 The gas occupies 2 L volume at STP. It is provided 300 Joule heat so that its volume becomes 2.5 L at 1 atm. Change in its internal energy will be
 1) 239 J 2) 205 J 3) 249.37 J 4) 220.37 J
- 24 If bond energies of H-H, Br-Br and H-Br are 433, 192 and 304 KJ mol^{-1} respectively. ΔH° for the reaction $\text{H}_{2(g)} + \text{Br}_{2(g)} \rightarrow 2\text{HBr}_{(g)}$ is
 1) -261 KJ 2) +103 KJ 3) 261 KJ 4) -103 KJ
- 25 Two moles of an ideal gas is expanded isothermally and reversibly from 1 L to 10 L at 300 K. The enthalpy of change (in KJ) for the process
 1) 11.4 KJ 2) -11.4 KJ 3) 0 KJ 4) 4.8 KJ
- 26 For a certain process $\Delta H = 280 \text{ KJ}$ and $\Delta S = 140 \text{ JK}^{-1} \text{ mol}^{-1}$ what is the minimum temperature at which the process will be spontaneous?
 1) 2000 K 2) 200 K 3) 140 K 4) 420 K
- 27 The signs of ΔH , ΔS and ΔG for a non spontaneous reaction at all temperatures would be
 1) +, +, - 2) +, -, + 3) -, -, - 4) +, +, +
- 28 In monatomic gases, ratio of specific heat at constant pressure to that of constant volume is
 1) $\frac{3}{5}$ 2) $\frac{5}{3}$ 3) $\frac{7}{5}$ 4) $\frac{4}{5}$
- 29 Following reaction occurs at 25° C $2 \text{ NO (g, } 1 \times 10^{-5} \text{ atm)} + \text{Cl}_2 \text{ (g, } 1 \times 10^{-2} \text{ atm)} \rightleftharpoons 2 \text{ NOCl (g, } 1 \times 10^{-2} \text{ atm)}$

- ΔG° is
- 1) -45.65 RJ 2) -66.53 RJ 3) -22.82 RJ 4) -35.65 RJ
- 30 Which one of the following has the maximum entropy of vaporization?
 1) water 2) Toluene (l) 3) Diethylether (l) 4) Acetone (l)
- 31 $1*1!+2*2! + \dots + n*n! =$
 1) $(n+1)! - 1$ 2) $(n-1)! + 1$ 3) $(n+1)! + 1$ 4) $(n-1)! - 1$
- 32 $1.2.3 + 2.3.4 + 3.4.5 + \dots$. n terms
 1) $n(n+1)(n+2)(3n+5) / 12$ 2) $n(n+1)(n+2)(n+3) / 4$
 3) $2n(n+1)(n+2)(n+3)$ 4) $n(n+1)(n+2)(3n+1) / 12$
- 33 $\frac{1^3}{1} + \frac{1^3+2^3}{1+3} + \frac{1^3+2^3+3^3}{1+3+5} + \dots$ n terms
 1) $n(2n^2+9n+13) / 24$ 2) $n(2n^3+9n+13) / 8$ 3) $n(n^2+9n+B) / 24$ 4) None
- 34 $\cos \theta \cos 2\theta \cos 4\theta \dots \cos 2^{n-1} \theta =$
 1) $\frac{\sin 2^n \theta}{2^n \sin \theta}$ 2) $\frac{\sin 2^n \theta}{\sin \theta}$ 3) $\frac{\cos 2^n \theta}{2^n \cos \theta}$ 4) $\frac{\cos 2^n \theta}{2^n \sin \theta}$
- 35 $7^{2n} + 3^{n-1} 2^{2n-3}$ is divisible by
 1) 7 2) 9 3) 25 4) 26
- 36 If $AB=A, BA=B$, then $A^2+B^2=$
 1) $A+B$ 2) $A-B$ 3) AB 4) 0
- 37 If $A = \begin{bmatrix} -8 & 5 \\ 2 & 4 \end{bmatrix}$ satisfies the equation $x^2+4x-p=0$, then p is
 1) 64 2) 42 3) 36 4) 24
- 38 If $A = \begin{bmatrix} 1 & 0 \\ 1 & 1 \end{bmatrix}, I = \begin{bmatrix} 1 & 0 \\ 0 & 1 \end{bmatrix}$ then which of the following holds for $n \geq 1$ by the principles of mathematical induction
 1) $A^n = nA - (n-1)I$ 2) $A^n = 2^{n-1} A - (n-1)I$ 3) $A^n = nA + (n-1)I$ 4) $A^n = 2^{n-1} A + (n-1)I$
- 39 If $A = \begin{bmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{bmatrix}$, then $\lim_{n \rightarrow \infty} \frac{1}{n} A^n$ is
 1) a null matrix 2) an identity matrix 3) $\begin{bmatrix} 0 & 1 \\ -1 & 0 \end{bmatrix}$ 4) none of these
- 40 If $A(\alpha) = \begin{bmatrix} \cos \alpha & \sin \alpha \\ -\sin \alpha & \cos \alpha \end{bmatrix}$, then $A(\alpha) A(\beta) =$
 1) $A(\alpha) - A(\beta)$ 2) $A(\alpha) + A(\beta)$ 3) $A(\alpha - \beta)$ 4) $A(\alpha + \beta)$
- 41 If A and B are square matrices of size $n \times n$ such that $A^2 - B^2 = (A-B)(A+B)$; then which of the following will be always true
 1) either of A or B is a zero matrix 2) either of A or B is an identity matrix
 3) $A = B$ 4) $AB = BA$
- 42 If $A = \begin{bmatrix} \cos \theta & \sin \theta & 0 \\ -\sin \theta & \cos \theta & 0 \\ 0 & 0 & 1 \end{bmatrix}$, then
 1) $AA^T = A^T A = I$ 2) $AA^T = A^T A = 0$ 3) $AA^T = A^T A = -I$ 4) none
- 43 If $A = \begin{bmatrix} 2 & 1 & -2 \\ a & 2 & b \end{bmatrix}$ is a matrix satisfying the equation $AA^T = 9I$, then (a,b) is
 1) (2,-1) 2) (-2,1) 3) (2,1) 4) (-2,-1)
- 44 If A,B are symmetric matrices of the same order then $AB-BA$ is
 1) a symmetric matrix 2) skew symmetric matrix 3) diagonal matrix 4) none

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Express $A = \begin{bmatrix} 2 & 0 & -3 \\ 4 & 3 & 1 \\ -5 & 7 & 2 \end{bmatrix}$ as a sum of symmetric and skew symmetric matrices

1) $\begin{bmatrix} 2 & 2 & -4 \\ 2 & 3 & 4 \\ -4 & 4 & 2 \end{bmatrix} + \begin{bmatrix} 0 & -2 & 1 \\ 2 & 0 & -3 \\ -1 & 3 & 0 \end{bmatrix}$

2) $\begin{bmatrix} 1 & 0 & -1 \\ 2 & 1 & 1 \\ -3 & 5 & 1 \end{bmatrix} + \begin{bmatrix} 1 & 0 & -2 \\ 2 & 2 & 0 \\ -2 & 2 & 1 \end{bmatrix}$

3) $\begin{bmatrix} 0 & 0 & -3 \\ 4 & 0 & 1 \\ -5 & 7 & 0 \end{bmatrix} + \begin{bmatrix} 2 & 0 & 0 \\ 0 & 3 & 0 \\ 0 & 0 & 2 \end{bmatrix}$

4) None

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ANSWER KEY

1	4	16	3	31	1
2	1	17	2	32	2
3	2	18	4	33	1
4	3	19	3	34	1
5	3	20	3	35	3
6	3	21	2	36	1
7	1	22	2	37	2
8	3	23	3	38	1
9	2	24	4	39	1
10	4	25	3	40	4
11	3	26	1	41	4
12	2	27	2	42	1
13	2	28	2	43	4
14	4	29	1	44	2
15	1	30	1	45	1

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