#### **MODULE – IV**

### **FUELS AND COMBUSTION**

- **4.4 Gaseous fuels**
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#### 4.4 Gaseous fuels

#### 4.4.1 Natural Gas

- Natural gas is obtained from wells dug in the oil bearing regions during the mining of petroleum.
- > Natural gas containing low molecular weight hydrocarbons is called dry or lean gas.
- > Natural gas containing high molecular weight hydrocarbons is called wet or rich gas.
- $\triangleright$  Calorific value ranges from 12,000 14,000 kcal / m<sup>3</sup>.

#### Composition of natural gas

Methane - 88.5 %

Ethane - 5.5%

**Propane - 3.7 %** 

Butane -1.8%

#### Uses

It is used as a domestic fuel.

It is used in the manufacture of number of chemicals.

It is used as a raw material for the manufacture of carbon black, methanol, etc.

## 4.4.3 Compressed natural gas (CNG) or Marsh gas

- ➤ Natural gas is compressed to high pressure of 1000 atm or cooled to 160°C to get compressed natural gas.
- ➤ It can be stored in steel cylinders.
- > It is a less polluting fuel.
- During combustion, there is no evolution of sulphur and nitrogen gases. It is a better fuel than petrol and diesel for automobiles.

#### Composition

Methane-88.5%

Ethane -5.5%

Propane -3.7%

Butane -1.8%

Pentane -0.5%

#### **Properties**

- i) It is a colourless gas.
- ii) It is a safer fuel.
- iii) It's ignition temperature is 550°C (higher temperature than gasoline and diesel).

- iv) It mixes with air easily.
- v) It leads to lesser emission than gasoline.

#### Uses

- i) It is an excellent domestic fuel that can be transported through pipes.
- ii) It is used as a fuel in thermal power plants for generating electricity.
- iii) It is used as a source of hydrogen gas in fertilizer industries.
- iv) It is used as an alternative to petrol and diesel for transport of vehicles.

#### 4.4.3 Liquefied petroleum gas (LPG)

- It is obtained as one of the top fractions in the fractional distillation of petroleum.
- It can be easily liquefied under pressure & stored in cylinders.
- It is a mixture of propane and butane.
- The composition is,

n-butane -38.5%

iso-butane - 37 %

**Propane** - 24.5%

Butylene and ethane - rest

## • Its calorific value is 27,800 Kcal /m<sup>3</sup>.

#### Characteristics

- ➤ It is a colourless gas stored in metallic cylinders in liquid state.
- > It is less polluting.
- > The products of combustion are non toxic.
- > It does not give out ash or smoke on burning.
- ➤ It has a special odour due to the addition of methyl mercaptan for safety purpose.
- > It burns cleanly without leaving any residue.
- > It has higher calorific value than coal gas & natural gas.
- ➤ It has high thermal efficiency.
- > It needs little care for maintenance.

#### Uses

- It is used as cooking gas.
- It is used as a heating source in hotels, bakeries, ang many other industries.
- It is used as a motor fuel.

#### 4.4.4 Power Alcohol

Power alcohol is a mixture of 75-80 % petrol, 20-25 % ethanol, and minute amount of **aromatic** compounds such as benzene. It can be used as a fuel for internal combustion engines. It is used as a very good fuel in motors.

#### Production of power alcohol

It involves two stages. They are as follows,

#### 1. Synthesis of ethyl alcohol

Ethyl alcohol can be synthesized by the fermentation of carbohydrates. This fermentation leaves only about 20% alcohol.

C6H12O6 Yeast 2C2H5OH + 2CO2

Glucose ethyl alcohol

The concentration of the obtained alcohol can be increased up to 97.6% by fractional distillation which is called rectified sprit.

- For the conversion of 97.6% of ethyl alcohol to absolute alcohol (100%) last traces of water must be removed.
- This can be done by distilling rectified spirit with benzene.

#### 2. Production of power alcohol from ethyl alcohol.

• Finally absolute alcohol is mixed with petrol at a concentration of 5 – 25% to get power alcohol.

#### **Properties**

- i) Power alcohol has a lower calorific value (7000 cal / g).
- ii) It has high octane number (90).
- iii) It's anti knocking properties are good.
- iv) It generates 10% more power than the gasoline of small quantity.
- v) It's compression ratio is also higher.

#### **Advantages**

- Power alcohol is cheaper than petrol.
- Alcohol has property of absorbing any traces of water if present in petrol.
- Ethyl alcohol contains 'O' atoms, which helps for complete combustion of power alcohol and the polluting emissions of CO, hydrocarbon, particulates are reduced largely.

#### **Disadvantages**

- Power alcohol has calorific value 7000 cal/gm much lower than calorific value of petrol 11500cal/gm. So use of power alcohol reduces power output up to 35%.
- Ethyl alcohol may undergo oxidation reaction to form acetic acid, which corrodes engine parts.
- As it contains 'O' atoms, the amount of air required for complete combustion of power alcohol is lesser and therefore carburetor and engine need to be modified.
- Due to high surface tension, it causes starting trouble in motors.

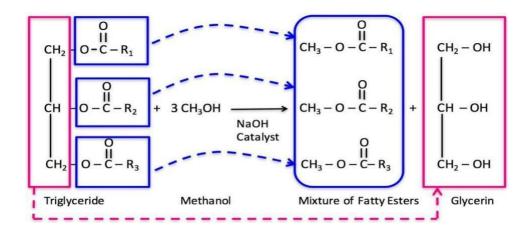
#### 4.4.5 BIODIESEL

#### **Definition & Explanation**

- > Biodiesel is a renewable, clean-burning diesel.
- Biodiesel is defined as mono-alkyl esters of long chain fatty acids derived from vegetable oils or animal fats which conform to ASTM D6751 specifications for use in diesel engines.
- Biodiesel can be used alone, or blended with petro-diesel in any proportions.
- ➤ Biodiesel can also be used as a low carbon alternative to heating oil.

#### Making of biodiesel

Biodiesel is made through a chemical process called trans - esterification. The process leaves behind two products -- methyl esters (biodiesel) and glycerine (a valuable byproduct usually sold to be used in soaps and other products).



#### **Advantages**

- > Biodiesel is environment friendly because it is made from renewable resources.
- ➤ It has lower emissions compared to petroleum diesel.
- > It is less toxic than table salt and biodegrades as fast as sugar.

- ➤ It is produced domestically from natural resources. So it is bio degradable.
- > It's use decreases our dependence on imported fuel and contributes to our own economy.

#### **Disadvantages**

- ➤ It gels during cold weather.
- > It absorbs water from atmosphere.
- > It decreases the efficiency of the engine.
- > It emits about 10% higher nitrogen oxides than conventional petroleum.

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